



# DIRECTORATE OF SCHOOL EDUCATION TAMILNADU

<b>12NPCB11 (2023-24)</b>	<b>NEET PRACTICE QUESTIONS (TEST-11)</b>	<b>Class : XII Time : 1.15 hrs Total Marks : 240</b>
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## Answer key

### 12th - CHEMISTRY

16.Ans: A)

In acidic medium the reaction is  $MnO_4^- + 8H^+ + 5e^- \rightarrow Mn^{2+} + 4H_2O$   
change in oxidation states is  $7 - 2 = 5$

17.Ans: D)

In This Bunsen burner and measuring cylinder is no use

18.Ans: B)

Molisch's test → Presence of carbohydrate

Barfoed test → presence of Mono saccharide

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A & B are carbohydrate

C- protein.

19.Ans : C)

Higher the  $R_f$  value lower the adsorption.

20. Ans: B)

$$N_1 V_1 = N_2 V_2$$

$$M_1 \times n_f \times V_1 = M_2 \times n_f \times v_2$$

$$0.05 \times 5 \times 10 = M_2 \times 2 \times 10$$

$$M_2 = \frac{0.05 \times 5}{2} = 0.125M$$

Strength of oxalic acid in g/L = Molarity × molar mass

$$g/L = 0.125 \times 126$$

$$= 15.75 \text{ g/L}$$

21. Ans : C)

No: of equivalents of  $\text{H}_2\text{SO}_4 = 100 \times 0.1 \times 2 = 20$

No. of equivalent &  $\text{NaOH} = 50 \times 0.1 \times 1 = 5$

No, of equivalents of  $\text{H}_2\text{SO}_4$  left =  $20 - 5 = 15$

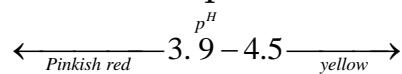
$$\text{Normality} = \frac{\text{No: of equivalent}}{\text{volume}}$$

$$\frac{15}{150} = \frac{1}{10} = 0.1N$$

22. Ans: C)

Weak base have  $p^{\text{H}}$  greater than 7. When methyl orange is added to a weak base, solution becomes yellow. If it titrated against strong acid, at the end  $p^{\text{H}}$  will be less than 3.1

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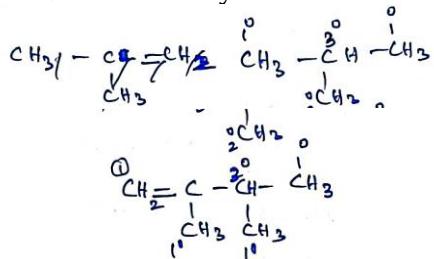
23. Ans: D) Pyrargyrite  $\text{Ag}_3\text{SbS}_3$

24. Ans : C)

Alkyl halide does not show immersions

- CN isomerism - NC
- CHO isomerism -  $\overset{\overset{\circ}{\text{C}}}{\underset{\parallel}{\text{O}}}^-$
- -OH isomerism - O-

25. Ans: 2, 3-dimethyl butane

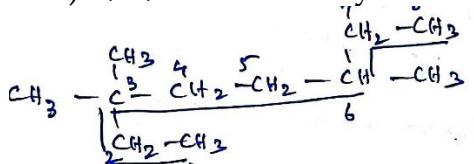


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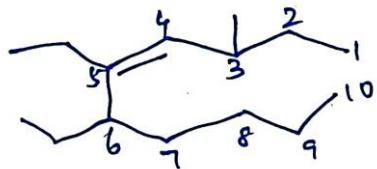


D) Propyne

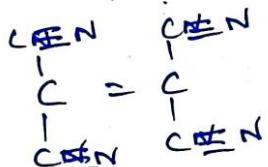
26.Ans: B) 3, 3, 6 - tri methyl octane



27.Ans: B) 5, 6 - diethyl -3-methyl dec-4-ene



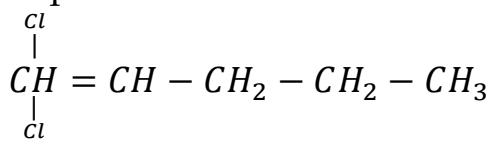
28.Ans: A)  $9\sigma$  and  $9\pi$



29.Ans: A) Correct decrying order

-COOH, -SO<sub>3</sub>H, -CONH<sub>2</sub>, -CHO

30.Ans: D) 1,1-dichloro-1-pentene



does not show geometrical isomerism



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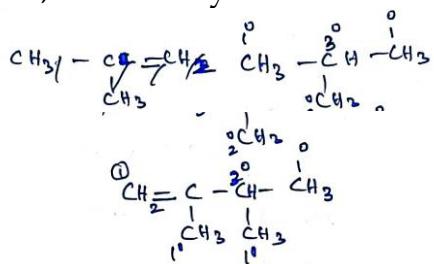
### 11th - CHEMISTRY

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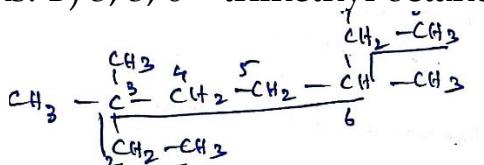


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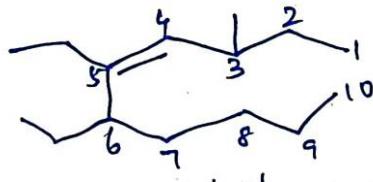


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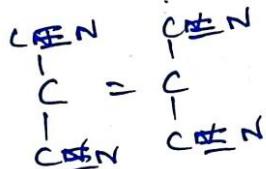
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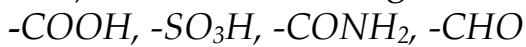
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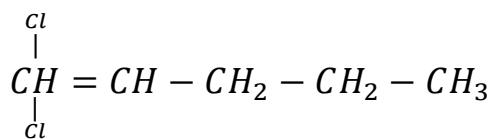
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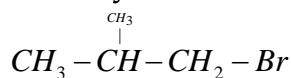


22.Ans: D) 1,1-dichloro-1-pentene



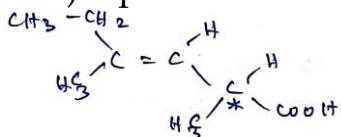
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23.Ans: B)  $\beta$ - bromo butyric acid

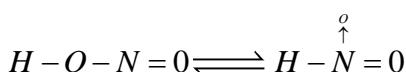
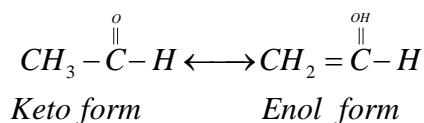


One chiral carbon So it is optically active

24.Ans: C) Optical isomerism and Geometrical isomerism



25.Ans: D) Ethanol

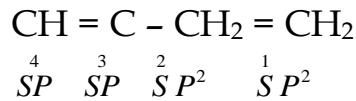


Nitsite

Nitro

26.Ans: C) iii. Pyridine    iv. Thiophene are heterocyclic compounds

27.Ans: B)  $sp^2, sp^2, sp, sp$

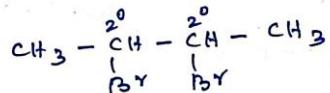


28.Ans: C) iii only

$\text{Br}-\text{CH}_2-\text{CH}=\text{CH}_2$  correct name is 3-bromo prop-1-ene

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30.Ans: B) 2, 2





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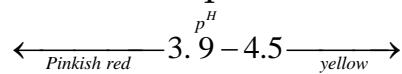
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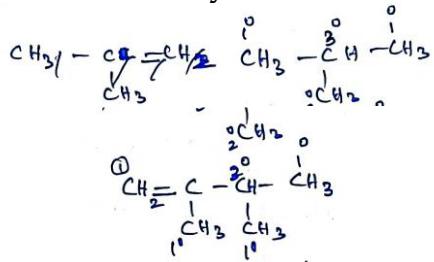
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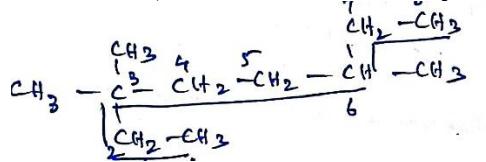


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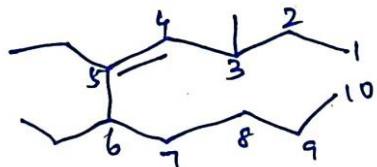
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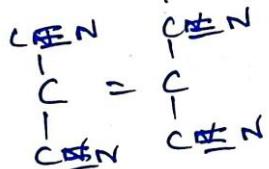
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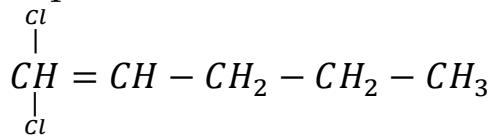
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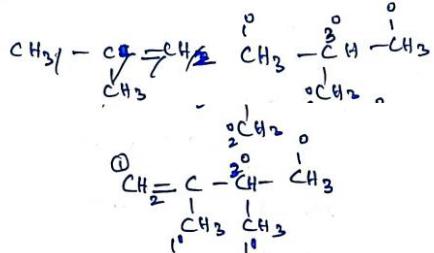
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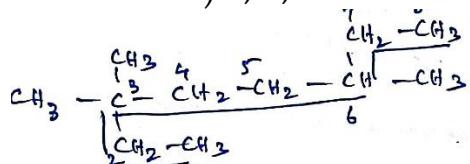


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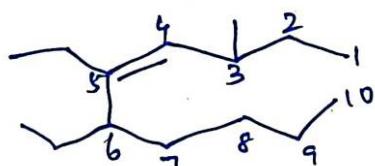


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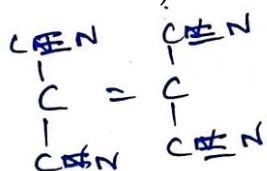
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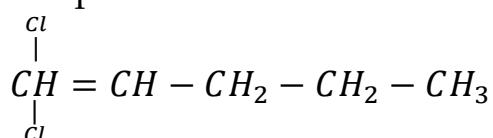


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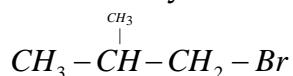
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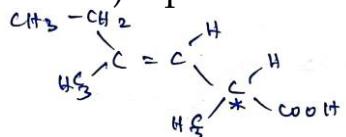
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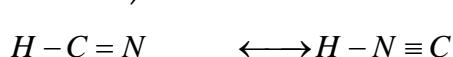


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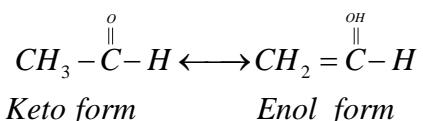
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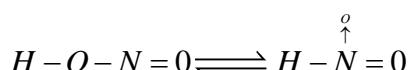


Hydrogen Cyanide      Hydrogenisocyange



Keto form

Enol form

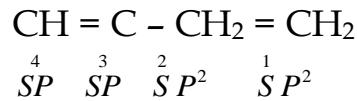


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